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## ORGANIC PREPARATIONS AND PROCEDURES INT. 7(2), 75-78 (1975)

#### A FACILE ROUTE TO

### BENZO[b]BICYCLO[3.2.2]NONENE DERIVATIVES

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The benzo[ $\underline{b}$ ]bicylco[3.2.2]nonene system (III) has not been studied extensively. Derivatives of this type (containing many substitutents) have been prepared by the Diels-Alder reaction<sup>1</sup> and by more complex reaction sequences.<sup>2</sup> The benzo-[ $\underline{b}$ ]bicyclo[3.2.2]nonene-4-yl acetate was formed in 24% yield during the acetolysis of 7-( $\beta$ -tosyloxyethyl)-benzo[ $\underline{a}$ ]cyclohepta-1,3-diene.<sup>3</sup> To our knowledge, besides this acetate and the corresponding alcohol and ketone (II) prepared from it<sup>3</sup>, no other simple benzo[ $\underline{b}$ ]bicyclo[3.2.2]nonene derivatives have yet been described.

We now report a convenient synthesis of the  $benzo[\underline{b}]$ bicyclo[3.2.2]nonene-4-one (II) by cyclization of 4-phenyl-



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cyclohexanecarboxylic acid (I) in the presence of polyphosphoric acid (PPA).

Evidently, only <u>cis</u>-4-phenylcyclohexanecarboxylic acid could be cyclized to the ketone II. The synthesis of <u>cis</u>-I is tedious<sup>4</sup> but it was pointed out<sup>4,5</sup> that vigorous acid treatment (conc. HCl, 185°) of <u>trans</u>-I (more easily available) afforded the <u>cis</u>-isomer in 10% yield. On this basis, we felt that the polyphosphoric acid (PPA) could simultaneously be used as an isomerization catalyst and as a cyclization agent.

The <u>trans</u>-4-phenylcyclohexanecarboxylic acid (<u>trans</u>-I) was prepared essentially by the Nenitzescu procedure<sup>6</sup> but the methyl 4-phenylcyclohexyl ketone was purified through the semicarbazone<sup>7</sup> and then oxidized<sup>8</sup> to <u>trans</u>-I.<sup>9</sup> PPA treatment of <u>trans</u>-I gave II in 66% yield, because the subsequent cyclization of <u>cis</u>-I continually shifted the equilibrium between the two isomeric acids towards the unstable <u>cis</u> form; if crude <u>trans</u>-I was used, the yield dropped to 60-62%. Wolff-Kishner reduction of ketone II afforded hydrocarbon III. The corresponding alcohol and acetate<sup>3</sup> could be readily synthesized from the ketone II by conventional methods.

# EXPERIMENTAL

<u>Benzo[b]bicyclo[3.2.2]nonen-4-one (II)</u>. A mixture of <u>trans-</u>4-phenylcyclohexanecarboxylic acid (3 g; 14.7 mmol) and PPA (prepared from 87 g of  $P_2O_5$  and 60 ml of 89%  $H_3PO_4$ , stirred at 100° for 1.5 hr) was heated with stirring at 130° for 3.5 hrs. The mixture was then poured into 100 ml ice-water and allowed to stand overnight. After extraction with 100 ml of ether and removal of the acidic material by washing with 5%

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NaOH, the solution was dried and evaporated to yield 1.8 g (66%) of a yellow oil. Purification by chromatography (30 g neutral alumina, eluent a mixture of 95% petroleum ether and  $\text{Et}_{20}$ ) afforded 1.65 (91%) II as a colorless liquid. IR (CCl<sub>4</sub>; cm<sup>-1</sup>): 1680 (vs), 2888 (s), 2961 (vs), 3040 (w), 3089 (w). NMR (CCl<sub>4</sub>;  $\delta$ ): 1.0-2.25 (8H, m, H<sub>6</sub>-H<sub>9</sub>), 2.91 (1H, m, H<sub>1</sub>), 3.18 (1H, m, H<sub>5</sub>), 7.1-7.55 (3H, m, H<sub>1</sub>',5',6'), 7.9-8.1 (1H, m, H<sub>4</sub>,). In the NMR spectrum of the complex Eu(DPM)<sub>3</sub> - II (molar ratio 0.281), H<sub>9</sub> as well as the aromatic proton H<sub>5</sub>, give rise to distinct signals.

<u>Oxime</u>, mp. 200-202° (MeOH). <u>Anal</u>. Calcd for C<sub>13</sub>H<sub>15</sub>NO: C, 77.57; H, 7.51; N, 6.96. Found: C, 77.19; H, 7.59; N, 6.90.

2,4-Dinitrophenylhydrazone, mp. 223-225° (dioxane-H<sub>2</sub>0). Anal. Calcd for  $C_{19}H_{18}N_4O_4$ : C, 62.28; H, 4.95; N, 15.29. Found: C, 62.41; H, 5.05; N, 15.10. UV-VIS (CHCl<sub>3</sub>,  $\lambda_{max}$  nm, log  $\varepsilon$ ): 384.9 (4.41).<sup>12</sup>

Benzo[b]bicyclo[3.2.2]nonene (III). Ketone II (0.3 g; 1.61 mmol), KOH (1.8 g; 32.2 mmol), 85% hydrazine (3.6 ml; 54 mmol) and diethylene glycol (18.1 ml) were refluxed for 2 hrs. The condenser was then set for distillation and the temperature raised to 200°; water was distilled, carrying some organic product. The mixture was heated at 200° for a further 3 hrs, then poured into 15 ml dil. HCl. The distillate was added and the mixture extracted with ether (50 ml). The ethereal extract was washed with water, dried (MgSO<sub>4</sub>) and evaporated to give a colorless oil (0.2 g; 72%). Purification (7 g alumina, eluent n-pentane) yielded 0.15 g (54%) of III.

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IR (CS<sub>2</sub> and CCl<sub>1</sub>; cm<sup>-1</sup>): 728 (s), 2865 (s), 2935 (vs), 3016 (w), 3060 (w). NMR (CCl<sub>4</sub>:  $\delta$ ): 1.4-2.05 (8H, m, H<sub>6</sub>-H<sub>0</sub>), 2.13 (1H, m, H<sub>5</sub>), 2.71 (1H, m,  $H_1$ ), 3.07 (2H, d,  $H_4$ ), 6.75-7.15 (4H, m,  $H_1$ ,  $H_4$ , 5', 6';  $J_{4,5} = 4$  Hz.

#### REFERENCES

- W. Davey and H. Gottfried, J. Org. Chem., 26, 3699 1. (1961).
- M. Takeshi and S. Akira, Bull. Chem. Soc. Japan, 34, 276 2. (1961).
- I. Dragutan, C. Draghici and E. Cioranescu, Rev. Roumaine 3. Chim., 20, 283 (1975).
- 4. H. E. Zimmerman and H. J. Giallombardo, J. Am. Chem. Soc., 78, 6259 (1956).
- 5.
- R. Rassow, Ann., <u>282</u>, 147 (1894). C. D. Nenitzescu and I. G. Gavat, ibid., <u>519</u>, 260 (1935). 6.
- 7. Methyl 4-phenylcyclohexyl kętone was isolated as a semicarbazone from the fraction<sup>6</sup> bp. 163-168°/12 mm (49% of this ketone, vpc analysis). Crystallization (EtOH) of the semicarbazone mixture afforded a derivative mp. 190-191° (lit.<sup>6,8</sup> 190-191° and 191°) which on acidic hydrolysis gave the pure ketone, mp. 53-55° (lit. 55°).
- 8. W. S. Johnson and R. D. Offenhauer, J. Am. Chem. Soc., 67, 1045 (1945).
- The resulted acid has mp. 208° (AcOH or MeOH) indicating 9. that it is the trans-isomer (lit.4,10 201-203° and 205-208°). Vpc analysis of its methylic ester showed a homogeneous product.
- 10. L. F. Fieser and M. T. Leffler, J. Am. Chem. Soc., 70, 3186 (1948).
- 11. Spectra were recorded with Zeiss Jena UR 20 (IR), a Specord UV-VIS (UV) and a Varian A-60A (NMR; TMS as internal standard). Vpc analyses were performed on a Carlo Erba model D; column: Carbowax 20 M (20% on Chromosorb W, 30-60 mesh), 2 m x 4 mm; 178°;  $H_2$  rate flow 80 ml/min.
- This maximum is consistent with a C=0 group included in 12. a seven-membered ring, for the DNPH of the related, perfectly conjugated, benzo[b]bicyclo[3.3.1]nonen-4-one displays a strong absorption at 390 nm (R. Huisgen and G. Seidl, Tetrahedron, 20, 231 (1964)).

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